

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: SH1096). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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## Glycoluril

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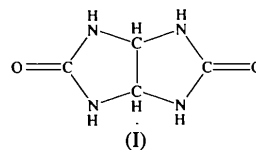
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## Abstract

The crystal structure of glycoluril [tetrahydroimidazo-[4,5-*d*]imidazole-2,5(1*H*,3*H*)-dione, C<sub>4</sub>H<sub>6</sub>N<sub>4</sub>O<sub>2</sub>] has been determined. The molecule has two equivalent planes containing urea moieties, and hence exhibits C<sub>2v</sub> symmetry. The dihedral angle is found to be 124.1 (4)°.

## Comment

Glycoluril (I) is a small molecule with relatively high symmetry and has a structure closely related to that of urea.



A recent study of the electronic spectrum of urea (Campbell & Clark, 1989) obtained experimental results that were contradictory to previous theoretical electronic structure calculations. In order to clarify the situation, polarized electronic spectra of single crystals of glycoluril have been measured (Xu & Clark, 1994). Interpretation of the polarized spectra requires a knowledge of the disposition of the molecules in the crystal. Although glycoluril has no important chemical applications except as a slow release fertilizer (Addiscott & Thomas, 1979; Shimizu, 1987), the fact that it is composed of two symmetrically disposed urea linkages is of significant value with respect to studies of the electronic structure of the basic chromophore. The structure of glycoluril has been investigated previously, but atomic co-

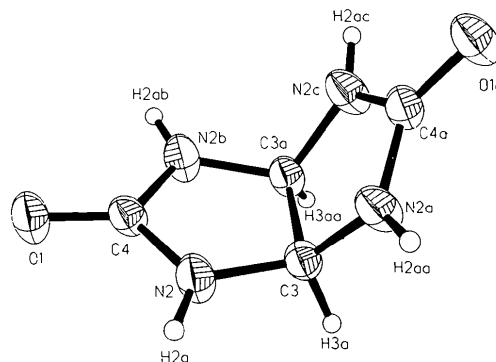


Fig. 1. A view of glycoluril with displacement ellipsoids at the 50% probability level. H atoms are drawn as small circles of arbitrary radii.

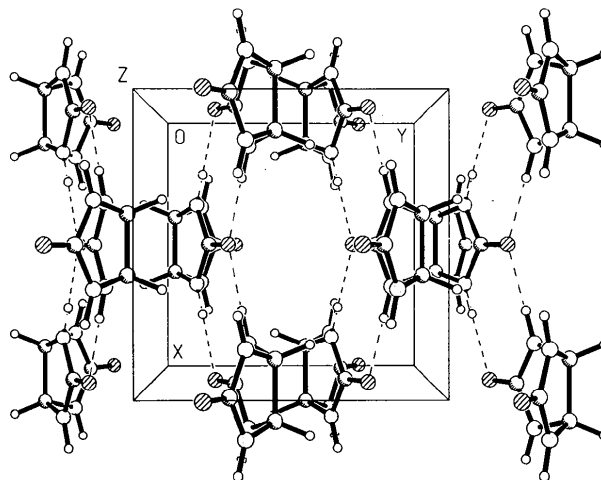


Fig. 2. A packing diagram viewed down the *c* axis showing the hydrogen bonding (O1...H2—N2 2.86 Å). The (010) face is determined to be a cleavage plane.

ordinates have not been published (Blessing & Detilta, 1979; Gilardi, 1982; Pierrot, Baldy, Wimmer, Gallo & Boileau, 1985).

## Experimental

Glycoluril was purchased from Aldrich and was used without further purification. Crystals were grown by the slow cooling of a filtered hot aqueous solution. The (010) face is predominant and shows perfect cleavage. This observation is consistent with the sheet-like arrangement of the structure (see Fig. 2).

### Crystal data

$C_4H_6N_4O_2$   
 $M_r = 142.12$   
 Orthorhombic  
*Cmcm*  
 $a = 7.373$  (2) Å  
 $b = 7.546$  (3) Å  
 $c = 10.579$  (3) Å  
 $V = 588.6$  (3) Å<sup>3</sup>  
 $Z = 4$   
 $D_x = 1.604$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation  
 $\lambda = 0.71073$  Å  
 Cell parameters from 16 reflections  
 $\theta = 4-13.5^\circ$   
 $\mu = 0.131$  mm<sup>-1</sup>  
 $T = 299$  K  
 Prism  
 $1.0 \times 0.4 \times 0.4$  mm  
 Colorless

### Data collection

Siemens R3m/V diffractometer  
 Wyckoff scans  
 Absorption correction: none  
 550 measured reflections  
 486 independent reflections  
 403 observed reflections  
 $[I > 2\sigma(I)]$

$R_{int} = 0.0165$   
 $\theta_{max} = 29.94^\circ$   
 $h = -1 \rightarrow 10$   
 $k = 0 \rightarrow 10$   
 $l = -14 \rightarrow 0$   
 3 standard reflections monitored every 197 reflections  
 intensity variation: 0.35%

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.0405$   
 $wR(F^2) = 0.1163$   
 $S = 1.114$   
 486 reflections  
 31 parameters  
 Only H-atom  $U$ 's refined  
 $w = 1/[\sigma^2(F_o^2) + (0.0785P)^2 + 0.1473P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = 0.001$   
 $\Delta\rho_{max} = 0.268$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.264$  e Å<sup>-3</sup>  
 Extinction correction: none  
 Atomic scattering factors from *International Tables for Crystallography* (1992), Vol. C, Tables 4.2.6.8 and 6.1.1.4)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

$$U_{eq} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i \cdot a_j$$

	x	y	z	$U_{eq}$
O1	0	0.2245 (2)	0.46658 (11)	0.0393 (4)
N2	0.14924 (12)	0.3510 (2)	0.63534 (9)	0.0358 (3)
C3	0.1059 (2)	0.4450 (2)	3/4	0.0274 (4)
C4	0	0.3016 (2)	0.57014 (15)	0.0285 (4)

Table 2. Selected geometric parameters (Å, °)

O1—C4	1.241 (2)	C3—C3 <sup>i</sup>	1.561 (3)
N2—C4	1.3512 (13)	C4—N2 <sup>ii</sup>	1.3512 (13)
N2—C3	1.4409 (13)		

C4—N2—C3	112.64 (11)	O1—C4—N2	125.47 (7)
N2 <sup>iii</sup> —C3—N2	114.67 (13)	N2—C4—N2 <sup>ii</sup>	109.05 (14)
N2 <sup>iii</sup> —C3—C3 <sup>i</sup>	102.82 (7)		

Symmetry codes: (i)  $-x, y, \frac{1}{2} - z$ ; (ii)  $-x, y, z$ ; (iii)  $x, y, \frac{1}{2} - z$ .

Data collection, cell refinement and data reduction: *SHELXTL-Plus* (Sheldrick, 1990a). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990b). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1994). Molecular graphics: *SHELXTL-Plus*.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and bond distances and angles involving non-H atoms have been deposited with the IUCr (Reference: KA1088). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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## A 3-Phenylsulfinyl-1-phenylsulfonylindole

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### Abstract

The title compound 2-(3,4-dihydro-2H-pyrrol-5-yl)-1-phenylsulfonylindol-3-yl phenyl sulfoxide,  $C_{24}H_{20}N_2O_3S_2$ , was isolated, together with 3-chloro-2-(3,4-dihydro-2H-pyrrol-5-yl)-1-phenylsulfonylindole, from the treatment of 2-(3,4-dihydro-2H-pyrrol-5-yl)indole with sodium hydride and phenylsulfonylchloride, showing that the latter can act as an electrophilic chlorinating agent and as a source of an electrophilic phenylsulfinylating species.